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Novel non-covalent supramolecular systems based on zinc(II) bis (dipyrromethenate)s with fullerenes



Alexander A. Ksenofontov ^{a,*}, Nataliya G. Bichan ^a, Ilya A. Khodov ^{a,b}, Elena V. Antina ^a, Mikhail B. Berezin ^a, Anatoly I. Vyugin ^a

^a GA. Krestov Institute of Solution Chemistry of the Russian Academy of Sciences, 1 Akademicheskaya Street, 153045 Ivanovo, Russia ^b Institute of Physics, Kazan Federal University, 18 Kremlyovskaya Street, 420008 Kazan, Russia

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ABSTRACT

This paper presents the results of UV/vis, fluorescence, FT-IR, DOSY, and DFT study of a novel non-covalent supramolecular systems based on zinc(II) bis(dipyrromethenate)s ($[Zn_2L_2]$) with C_{60} . It was found that zinc(II) bis (dipyrromethenate) forms stable supramolecular π - π -complexes ($[Zn_2L_2(C_{60})_4]$) with C_{60} . DFT calculations revealed stable complexation between $[Zn_2L_2]$ and C_{60} . The $[Zn_2L_2(C_{60})_4]$ LUMO energy levels are predominantly spread on the C_{60} unit and the HOMO energy levels are mainly spread on the $[Zn_2L_2]$. The photoinduced electron transfer testing by the substantial fluorescence quenching of the $[Zn_2L_2]$ by non-covalent bonded C_{60} gives the positive result that shows prospects of the studying $[Zn_2L_2(C_{60})_4]$ as the active layers in solar energy conversion devices.

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1. Introduction

Fullerenes are remarkable electron acceptors. It is common to consider fullerenes are electron-deficient polyene systems rather than aromatic molecules. Extensive three-dimensional spherical form of fullerene molecules facilitates delocalization of charges along their entire structure and provides unique opportunities for stabilization of charged particles [1,2]. In addition, fullerenes possess unique electrochemical and characteristic spectral properties [3,4]. Special attention is attracted to the search and receipt of supramolecular systems based on various nature fullerenes [5,6]. This is due to the fact that complexes can be used to create photosynthetic and photonic devices [7–9]. Effective complexation with a molecular ligand is feature of supramolecular chemistry of fullerenes. Host-guest systems are one of the typical and often obtained supramolecular systems based on fullerenes. Various host molecules have been designed in recent past, for example, carbon nanotubes [10], crown ethers [11,12], calixarenes [13], cyclodextrins [14]. On the other hand, porphyrins [15–19] and phthalocyanines [20,21] are often used as receptors for fullerenes covalent binding. Porphyrin (phthalocyanine) - fullerene covalent systems are good platforms for creating alternative photosynthetic reaction centers. This is due to the high electron-donor ability of extended conjugated macrocyclic π -systems of porphyrins and phthalocyanines [22,23]. A large

number of papers are devoted to the studies of covalently bound porphyrin-fullerene complexes [24–26]. Their study was an important step in understanding the mechanism governing photoinduced electron transfer in molecular systems [23,27]. However, the supramolecular assembly of non-covalent fullerene-containing systems is another rapidly developing approach for the photoactive components production. This method has the advantages of synthesizing covalent porphyrinfullerene systems. Ease of implementation is one of the main advantages. Therefore, the search of powerful π -electron-donor systems capable of interacting with fullerene molecules due to π - π interactions is an important task. From this point of view, a new class of luminophores based on zinc(II) bis(dipyrromethenates) ($[Zn_2L_2]$) is of considerable interest. It is important to note that the data on supramolecular complexes of $[Zn_2L_2]$ with fullerenes are absent in modern literature. However, [Zn₂L₂] having intense luminescent properties are promising building blocks for creating molecular architectures with charge separation.

Binuclear helicates ([Zn₂L₂]) are luminescent dyes with intense and clear absorption spectra and high sensitivity of fluorescence characteristics to the medium properties. It was shown earlier that [Zn₂L₂] form stable supramolecular complexes of the composition [Zn₂L₂X_n] (X is an aromatic molecular ligand) due to π -stacking of aromatic systems of helicate dipyrromethene domains and X molecules [28–30]. The π -stacking efficiency, luminophore polarization, and components mobility in supramolecular complex increase the probability of nonradiative transitions in the excited state. Thus, creating of the capable of

^{*} Corresponding author. *E-mail address:* ivalex.09@mail.ru (A.A. Ksenofontov).