

## Two-phase titration of dialkyl dithiophosphoric acids based on their ion pair extraction

Garifzyanov A., Tel'nova Z., Savel'eva N., Toropova V., Budnikov G.

*Kazan Federal University, 420008, Kremlevskaya 18, Kazan, Russia*

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### Abstract

The distribution of ion pairs formed by a number of O,O-dialkyl dithiophosphoric acids and decylpyridinium (DP) in a water-chloroform two-phase system was studied. Extraction constants (KDP) were determined. A linear correlation between  $\log KDP$  and steric and hydrophobic constants of substituents at the phosphorus atom was found. Two-phase potentiometric titration can be applied to the selective determination of dithiophosphoric acids. © 1998 MAEe Cyrillic signK Hayka/Interperiodica Publishing.

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