



ELSEVIER

Available online at [www.sciencedirect.com](http://www.sciencedirect.com)

Nuclear Instruments and Methods in Physics Research B 206 (2003) 277–281

**NIM B**  
Beam Interactions  
with Materials & Atoms[www.elsevier.com/locate/nimb](http://www.elsevier.com/locate/nimb)

## Coloration of natural beryl by iron ion implantation

R.I. Khaibullin <sup>a,c,\*</sup>, O.N. Lopatin <sup>b</sup>, F.G. Vagizov <sup>b</sup>, V.V. Bazarov <sup>a</sup>,  
A.I. Bakhtin <sup>b</sup>, I.B. Khaibullin <sup>a</sup>, B. Aktas <sup>c</sup>

<sup>a</sup> Laboratory of Radiation Physics, Kazan Physical–Technical Institute, Sibirskij Trakt 10/7, 420029 Kazan, Russia

<sup>b</sup> Kazan State University, Kremlevskaya 18, 420008 Kazan, Russia

<sup>c</sup> Gebze Institute of Technology, P.K. 141 41400, Gebze/Kocaeli, Turkey

### Abstract

Natural colorless crystals of Ural beryl were implanted at room temperature with 40 keV Fe<sup>+</sup> ions with fluences in the range of  $0.5\text{--}1.5 \times 10^{17}$  ion/cm<sup>2</sup>. As-implanted samples show dark-grey tone due to radiation damage of beryl crystal. Subsequent thermal annealing of irradiated crystals in oxygen at 600 °C for 30 min results in the color change, to yellowish or yellow–orange tones with golden luster, depending on value of iron fluence. The nature of beryl coloration was studied by optical absorption, Mössbauer and Rutherford backscattering (RBS) spectroscopies. It was established that the thermal treatment of iron-irradiated beryl lead to inward diffusive redistribution of iron ions. An appearance of optical absorption bands connected with charge-transfers  $\text{O}^{2-} \rightarrow \text{Fe}_{\text{VI}}^{3+}$  and  $\text{O}^{2-} \rightarrow \text{Fe}_{\text{IV}}^{2+}$ ,  $\text{Fe}_{\text{IV}}^{3+}$  determine the yellow tone in colored beryls. Most of implanted iron ions are founded in both tetrahedral  $\text{Fe}_{\text{IV}}^{2+}$  and octahedral  $\text{Fe}_{\text{VI}}^{3+}$  sites where they may substitute beryllium and aluminum host ions by isomorphic way.

© 2003 Elsevier Science B.V. All rights reserved.

PACS: 61.72.Ww; 61.82.Ms; 91.60.Mk; 76.80.+y

Keywords: Ion implantation; Beryl; Optical absorption; Mössbauer spectroscopy

### 1. Introduction

Beryl,  $\text{Be}_3\text{Al}_2(\text{SiO}_3)_6$ , is a ring silicate mineral whose hexagonal crystalline structure was firstly described by Bragg et al. [1]. Pure beryl is colorless and transparent. However, beryl crystals may show various color tones if they contain transition metal ions as an impurity localized in crystalline

sites [2,3]. For example, green color of emerald is caused by  $\text{Cr}^{3+}$  ions in aluminum sites, and pink beryl (morganite) is colored by  $\text{Mn}_{\text{Al}}^{3+}$  ions. Different correlation of the  $\text{Fe}^{2+}$  and  $\text{Fe}^{3+}$  ions in structural and interstitial positions of the crystal determine the color of the Fe-containing beryls from blue aquamarines to yellow–orange heliodors. Note that intensively colored yellow beryls with metallic luster are known in gemology as golden beryls.

Ion beam technology enables to introduce any chemical element in the near-surface layer of materials, and a possibility of coloring natural minerals by ion implantation may be an interesting issue. However, few works (e.g. [4,5]) have been

\* Corresponding author. Address: Laboratory of Radiation Physics, Kazan Physical–Technical Institute, Sibirskij Trakt 10/7, 420029 Kazan, Russia. Tel.: +7-8432-761241/721241; fax: +7-8432-765075/725075.

E-mail address: [rik@kfti.knc.ru](mailto:rik@kfti.knc.ru) (R.I. Khaibullin).